Synthesis and *In Vitro* Study of a New Class of Methylenebis-4,6-diarylbenzo[*d*]isoxazoles as Potential Antifungal Agents

A. Srinivas, A. Nagaraj, and Ch. Sanjeeva Reddy*

Department of Chemistry, Kakatiya University, Warangal 506 009, India *E-mail: chsrkuc@yahoo.co.in Received June 26, 2008 DOI 10.1002/jhet.100 Published online 26 May 2009 in Wiley InterScience (www.interscience.wiley.com).



A new class of methylene-bis-4,6-diarylbenzo[*d*]isoxazoles **8a–i** was synthesized by the reaction of methylene-bis-aryl-6-hydroxymethylene-2-cyclohexenone **6** with hydroxylamine hydrochloride, followed by aromatization with DDQ. Chemical structures of the newly synthesized compounds were elucidated by their IR, ¹H NMR, ¹³C NMR, MS, and elemental analyses. Furthermore, all the compounds were screened for their antifungal activity against various fungi and compared with their monomeric compounds. Among the synthesized compounds, **8b**, **8g**, and **8i** were found to be the most active against *Candida albicans* (ATCC 10231), *Aspergillus fumigatus* (HIC 6094), *Trichophyton rubrum* (IFO 9185), and *Trichophyton mentagrophytes* (IFO 40996). It is interesting to note that the compounds **8a**, **8g**, and **8i** showed fungicidal activity toward *C. albicans* at the concentration of 3.12 µg/mL, which is less than the concentration of standard Amphotericin B.

J. Heterocyclic Chem., 46, 497 (2009).

INTRODUCTION

The treatment of infectious diseases still remains an important and challenging problem because of a combination of factors including emerging infectious diseases and the increasing number of multidrug resistant microbial pathogens. In spite of a large number of antibiotics and chemotherapeutics available for medical use, the emergence of old and new antibiotic resistance created in the last decades revealed a substantial medical need for new classes of antimicrobial agents. There is a real perceived need for the discovery of new compounds endowed with antimicrobial activity, which are distinct from those of well known classes of antibacterial agents to which many clinically relevant pathogens are now resistant. Similarly in recent decades, an increased incidence of fungal infections has been observed as a consequence of the growing number of immunocompromised patients and the frequent use of antibacterial and cytotoxic drugs. For many fungal infections, polyenes, such as Amphotericin B, represent the standard therapy. Polyenes bind to membrane sterols, leading to membrane permeability, leakage, and cell death. However, the clinical use of Amphotericin B is limited by a high frequency of renal toxicity and several adverse effects [1]. Although the various molecules designed and synthesized for the above aim and to reduce the adverse effects, it was demonstrated that isoxazole derivatives could be considered as possible antifungal agents [2]. The other activities include, inhibition of AB precursor protein (APP) [3], inhibition of protein tyrosine phosphatase 1B [4], antiviral [5], anthelmintics [6], antiinflammatory [7], anticonvulsant [8], insecticidal [9], antitubercular [10], immunomodulatory [11], and hypolipemics [12]. Valdecoxib, an isoxazole derivative, is now widely used in the market as an anti-inflammatory drug [13]. The most general and widely employed synthetic route to isoxazoles involves reaction of chalcones with hydroxylamine hydrochloride [14] or 1,3-dipolar cycloaddition of nitrile oxides to alkynes [15] or condensation of open chain α -hydroxymethylene ketones with hydroxylamine [16] or from 3,5-diarylcyclohexenone and hydroxylamine [17].

Following the successful introduction of antimicrobial agents, inspired by the biological profile of isoxazoles and their increasing importance in pharmaceutical and biological fields, and in continuation of our research on biologically active heterocycles [18], and to enhance the biological activity of isoxazole derivatives, it was considered worth while to synthesize certain new chemical entities incorporating two active pharmacophores in a single molecular framework. In this article, we wish to report the synthesis of a new class of methylene-bis-4,6-diarylbenzo[*d*]isoxazoles **8** in good yields from methylene-bis-aryl-6-hydroxy-methylene-2-cyclohexenone **6** (Scheme 1) and *in vitro* antifungal activity.

RESULTS AND DISCUSSION

The key intermediate, 6 required for the synthesis of the title compounds was prepared according to the procedure outlined in the Scheme 1. Condensation of the salicylaldehyde 1 and trioxane in the presence of a mixture of conc. sulfuric acid and acetic acid gave methylene-bis-salicylaldehyde 2 in good yield [19]. Compound 2 on reaction with the aromatic/heteroaromatic methyl ketones in the presence of alc. KOH at room temperature gave methylene-bis-chalcones 3 (yield over 90%) [18(g)], the reaction time as well as the product yield varies depending on the corresponding reagents. The crude product, contaminated by some starting materials, was purified by extracting with ether. Knoevenagel condensation of compound 3 with ethyl acetoacetate gave methylene-bis-aryl-6-carbethoxycyclohexenones 4 (yield over 80%). Decarboxylation of compound 4, in the presence of HCl/AcOH at reflux temperature, resulted in methylene-bis-arylcyclohexenone 5 (yield over 80%), which on Claisen-like condensation with ethylformate in the presence of sodium methoxide at room temperature afforded methylene-bis-aryl-6-hydroxymethylene-2cyclo-hexenone 6 in good yields. Compound 6 on

Scheme 1. Reagents and conditions: (i) trioxane, $H_2SO_4/AcOH$, reflux, 81%; (ii) ArCOCH₃, KOH/EtOH, rt, 82–95%; (iii) EAA, NaOEt/EtOH, reflux, 78–86%; (iv) HCl/AcOH, reflux, 74–82%; (v) HCOOEt, NaOMe/C₆H₆, rt, 79–88%; (vi) H₂NOH·HCl, AcOH, reflux, 77–86%; and (vii) DDQ/dry C₆H₆, N₂–atm, 76–83%.



Journal of Heterocyclic Chemistry DOI 10.1002/jhet

	Table 1 Synthesis of compounds 8a-i.				
Product ^a	Ar	Mol. formula	Yield (%) ^b	M.p.°C	
8a		$C_{39}H_{26}N_2O_4$	78	158-60	
8b	Br-	$C_{39}H_{24}Br_2N_2O_4$	84	171–72	
8c	H ₂ N-	$C_{39}H_{28}N_4O_4$	79	158–60	
8d	ci–	$C_{39}H_{24}Cl_2N_2O_4$	86	144-46	
8e	MeO-	$C_{41}H_{30}N_2O_6$	77	135–37	
8f		$C_{35}H_{22}N_2O_6$	80	162–63	
8g	N	$C_{33}H_{20}N_4O_4S_2$	82	149–50	
8h		$C_{37}H_{24}N_4O_4$	77	151–53	
8i		$C_{35}H_{22}N_6O_4$	81	162–64	

 ^a All the products were characterized by IR, NMR, MS, and elemental analyses.
 ^b Isolated yields after purification.

cyclocondensation with hydroxylamine hydrochloride in refluxing acetic acid gave dihydroisoxazole derivative 7, in good to excellent yields (yields over 80%). Subsequent aromatization of 7 with dichlorodicyanoparabenzoquinone (DDQ), under N₂-atomsphere at reflux temperature, gave methylene-bis-4,6-diarylbenzo[*d*]isoxazoles 8 in good yields (Table 1). In the ¹H NMR spectra of 8, the absence of signals corresponding to methine and methylene protons of cyclohexadiene ring indicates that aromatization has indeed taken place. The structure of all the synthesized compounds was confirmed by their IR, ¹H NMR, ¹³C NMR, MS, and elemental analyses. Furthermore, the compounds was subjected to antifungal testing and compared with their monomeric compounds prepared from the literature method [17].

In this work, a series of nine new benzo-[d]isoxazole derivatives were synthesized and the Scheme 1 illustrates the strategy used for the synthesis of target com-

pounds. Salicylaldehyde and trioxane was used as starting materials to prepare isoxazole derivatives. In the IR spectra of compounds 8a-i, the C=N and N-O bands of the isoxazole moiety was observed at about 1600 and 1470 cm⁻¹, respectively. In the ¹H NMR spectra of compounds 8a-i recorded in DMSO- d_6 , the signal due to methylene bridge protons appeared at 3.98-4.06 ppm as a singlet and the N=CH proton appeared at 8.20-8.28 ppm as a singlet, proved that these compounds have isoxazole nucleus. All the other aromatic and aliphatic protons of 8a-i were observed at the expected regions. In the ${}^{13}C$ NMR spectra of compounds 8a-i that are taken in DMSO- d_6 , the prominent signal corresponding to C-3, C-3a, and C-7a observed at 150.3, 128.0-120.5, 173.0-170.0 ppm, respectively, have proved for further evidence for their structures. Mass spectra of all the synthesized compounds showed M⁺/ M^++1 peaks are in agreement with their molecular formulae.

Antifungal activity. The newly prepared compounds were screened for their antifungal activity against four fungal organisms viz. Candida albicans (ATCC 10231), Aspergillus fumigatus (HIC 6094), Trichophyton rubrum (IFO 9185), and Trichophyton mentagrophytes (IFO 40996) by the broth dilution method, recommended by National Committee for Clinical Laboratory Standards (NCCLs) [20]. The C. albicans was grown for 48 h at 28°C in YPD broth (1% yeast extract, 2% peptone, and 2% dextrose), harvested by centrifugation, and then washed twice with sterile distilled water. A. fumigatus, T. rubrum, and T. mentagrophytes were plated in potato dextrose agar (PDA) (Difco) and incubated at 28°C for 2 weeks. Spores were washed three times with sterile distilled water and resuspended in distilled water to obtain an initial inoculum size of 10⁵ spores/mL. Each test compound was dissolved in DMSO and diluted with potato dextrose broth (Difco) to prepare serial twofold dilutions in the range 100 to 0.8 µg/mL. Ten microliters of the broth containing about 10^3 (for yeast) and 10^4 (for filamentous fungi) cells/mL of test fungi was added to each well of a 96-well microtiter plate. Culture plates were incubated for $\sim 48-72$ h at 28° C. The antifungal activity of each compound was compared with the standard drug Amphotericin B. Minimum inhibitory concentration (MIC, µg/mL) was measured and compared with controls; the MIC values of the compounds screened are presented in Table 2.

The antifungal screening data showed only moderate activity of the tested compounds. Among the screened compounds, **8b**, **8g**, and **8i** in which isoxazole moiety bearing 4-bromophenyl, 5-thiazole, and 2-pyrazine nucleus, respectively, were showed high activity against all the microorganisms employed. The activities of these three compounds are almost equal to the standard. It is

Compd.	Minimum inhibitory concentration (MIC, µg/mL)				
	Candida albicans	Aspergillus fumigatus	Trichophton rubrum	Trichophyton mentagrophytes	
8a	25.0 ^a (50.0) ^b	50.0 ^a (—) ^b	$25.0^{a} (-)^{b}$	50.0 ^a (—) ^b	
8b	3.12 (12.5)	3.12 (12.5)	3.12 (12.5)	3.12 (12.5)	
8c	12.5 (12.5)	25.0 (25.0)	25.0 (50.0)	6.25 (12.5)	
8d	6.25 (25.0)	25.0 (50.0)	— (—)	12.5 (50.0)	
8e	12.5 (25.0)	12.5 (25.0)	6.25 ()	6.25 (25.0)	
8f	25.0 (50.0)	12.5 (25.0)	6.25 (25.0)	6.25 (25.0)	
8g	3.12 (25.0)	3.12 (12.5)	3.12 (6.25)	3.12 (12.5)	
8h	12.5 (50.0)	3.12 (6.25)	3.12 (25.0)	3.12 (6.25)	
8i	3.12 (25.0)	3.12 (12.5)	3.12 (12.5)	3.12 (12.5)	
aphotericin B	6.25	3.12	3.12	3.12	

 Table 2

 Antifungal activity of compounds 8a–i.

— Indicates fungi are resistant to the compound $>50 \ \mu\text{g/mL}$ conc.

^a Activity of dimeric compounds.

^b Activity of their monomeric compounds.

also interesting to note that the compounds **8b**, **8g**, and **8i** showed activity toward *C. albicans* at the concentration of 3.12 µg/mL, which is less than the concentration of the standard Amphotericin B. Similarly, compound **8h** in which isoxazole ring bearing 2-pyridine nucleus, also showed good antifungal activity equal to the activity of the standard but only toward *A. fumigatus* and *T. rubrum*. The remaining compounds showed moderate to good antifungal activity. Furthermore, the activity of dimeric compounds was compared with that of their monomeric compounds prepared from the reported method [17]. The results reveal that almost all the dimeric compounds showed enhanced activity than their monomeric compounds (Table 2).

In conclusion, a new class of methylene-bis-4,6-diarylbenzo[d]isoxazoles **8a–i** has been designed and synthesized. The antifungal activity of these compounds was evaluated against various fungi. Among the synthesized compounds, **8b**, **8g**, and **8i** showed good activity against test fungi and emerged as potential molecules for further development. With this set of analogues, we are now in a position to investigate the multiple biological activities for these compounds.

EXPERIMENTAL

All the chemicals and solvents were of analytical grade and used as purchased. A proper safety measure was taken while carring out the reactions. Evaporations were performed at reduced pressure below 40°C. The reactions and purifications were monitored by TLC on aluminium sheets coated with silica gel 60 F_{254} (Merck), column chromatography on silica gel 60 (Merck). Melting points were taken using a Fisher-Johns melting point instrument and are uncorrected. IR spectra were obtained on a Perkin-Elmer FTIR 5000 spectrometer, using KBr pellets. ¹H NMR and ¹³C NMR spectra were

obtained with Varian Gemini (¹H: 300 MHz, ¹³C: 75 MHz) spectrometer, and the chemical shifts were reported as parts per million (δ ppm) down field from internal tetramethylsilane and coupling constants (*J*) in Hz. Mass spectra were obtained on a VG Micromass 7070H spectrometer. Elemental analyses were performed on a Perkin-Elmer 240 CHN elemental analyzer.

Ethyl-6-(5-{3-[6-(ethoxycarbonyl)-5-oxo-3-phenyl-3-cyclohexenyl]-4-hydroxybenzyl}-2-hydroxyphenyl)-2-oxo-4-phenyl-3-cyclohexene-1-carboxylate (4a). To a solution of sodium metal (2 g) in ethanol (30 mL), a mixture of freshly distilled ethylacetoacetate (3.9 mL, 0.03 mol) and compound 3a (4.6 g, 0.01 mol) dissolved in ethanol (20 mL) was added. The resulting solution was refluxed on a water bath for 4 h. Allowing the reaction mixture to cool and crystallization of the formed precipitate from ethanol gave 4a (82% yield) as brown solid; mp 150–152°C; ¹H NMR (DMSO-d₆): δ 1.10 (6H, t, CH₃), 2.87 (4H, d, CH₂), 3.72 (2H, s, CH₂), 3.81 (2H, d, CH), 3.87 (2H, q, CH), 4.06 (4H, q, CH₂), 5.20 (2H, s, OH), 6.10 (2H, s, CH), 6.62 (2H, d, J = 9.2 Hz, ArH), 6.79 (2H, d, J = 9.2 Hz, ArH), 6.80 (2H, s, ArH), 7.10-7.14 (10H, m, ArH); ¹³C NMR (DMSO-*d*₆): δ 17.0, 30.7, 37.0, 42.1, 60.6, 61.2, 117.4, 121.9, 123.8, 125.5, 127.9, 128.2, 128.7, 130.0, 133.4, 142.3, 149.5, 154.6, 176.7, 190.1; IR (KBr): v 3452, 3065, 1702, 1695, 1597, 1245 cm⁻¹; MS: m/z 685 (M⁺ +1). The other compounds 4b-i were also prepared by the similar procedure.

5-{2-Hydroxy-5-[4-hydroxy-3-(5-oxo-3-phenyl-3-cyclohexenyl)benzyl]phenyl-3-phenyl}-2-cyclohexen-1-one (5a). To a mixture of glacial AcOH (100 mL) and conc. HCl (50 mL) was added compound **4a** (6.5 g, 0.01 mol) in portions. The mixture was heated to reflux for 10 h. After cooling to room temperature, the reaction mixture was concentrated *in vacuo*. The residue was taken up with ethyl acetate and washed with water and brine, dried over MgSO₄, filtered, and evaporated *in vacuo* to give oil, which soon solidified. It was purified by recrystallization from ethanol to give compound **5a** (79% yield) as brown solid; mp 132–134°C; ¹H NMR (DMSO-*d*₆): δ 2.67 (4H, d, CH₂), 2.70 (4H, d, CH₂), 3.72 (2H, s, CH₂), 3.83 (2H, m, CH), 5.20 (2H, s, OH), 6.10 (2H, s, CH), 6.62 (2H, d, *J* = 9.2 Hz, ArH), 6.82 (2H, s, ArH), 7.00 (2H, d, J = 9.2 Hz, ArH), 7.10– 7.14 (10H, m, ArH); ¹³C NMR (DMSO- d_6): δ 31.0, 38.9, 42.0, 47.5, 116.1, 122.6, 124.5, 126.2, 127.4, 127.8, 128.5, 131.2, 132.3, 140.5, 150.1, 155.2, 198.0; IR (KBr): υ 3357, 3025, 2932, 1687, 1596 cm⁻¹; MS: m/z 541 (M⁺ +1). The other compounds **5b–i** were also prepared by the similar procedure.

5-[2-Hydroxy-5-(4-hydroxy-3-{6-[(Z)-1-hydroxymethylidene]-5-oxo-3-phenyl-3-cyclohexenyl}benzyl)phenyl]-6-[(Z)-1-hydroxymethylidene]-3-phenyl-2-cyclohexen-1-one (6a). In a solution of 10% sodium methoxide (10 mL) in benzene (25 mL), ethylformate (2.24 mL, 0.03 mol) was added and then compound 5a (5.4 g, 0.01 mol), dissolved in benzene (10 mL), was added over 30 min. The resulting solution was stirred for 10 h at room temperature and allowed to stand overnight, then evaporated to dryness. The suspension obtained was mixed with cold water, acidified with dil HCl (20 mL), and extracted three times with ether (40 mL). The organic layer was dried over MgSO₄, evaporated in vacuo to give solid, and purified by crystallization in ethanol to afford pure 6a (81% yield) as yellow solid; mp 143–145°C; ¹H NMR (DMSO- d_6): δ 3.22 $(4H, d, CH_2), 3.72 (2H, s, CH_2), 4.12 (2H, t, CH), 5.67 (2H, s,$ CH), 6.40-6.49 (4H, m, ArH), 6.80 (2H, s, ArH), 7.10-7.14 (10H, m, ArH), 7.92 (2H, s, CH), 8.97 (2H, s, OH); ¹³C NMR (DMSO-d₆): § 37.8, 42.0, 44.1, 115.5, 116.7, 126.2, 127.6, 127.9, 128.0, 128.9, 130.2, 130.6, 131.1, 142.0, 149.7, 156.1, 167.6, 191.3; IR (KBr): v 3320, 3028, 2952, 1662, 1620, 1597 cm⁻¹; MS: m/z 596 (M⁺). The other compounds **6b–i** were also prepared by the similar procedure.

4-[4-Hydroxy-3-(6-phenyl-4,5-dihydrobenzo[d]isoxazol-4yl)benzyl]-2-(6-phenyl-4,5-dihydrobenzo[d]isoxazol-4-yl)phenol (7a). To a solution of 6a (5.9 g, 0.01 mol) in glacial acetic acid (50 mL), hydroxylamine hydrochloride (2.0 g, 0.03 mol) was added. After stirring at 80°C for 10 h, the mixture was concentrated in vacuo. To the residue was added water and twice extracted with ether. The organic layer was washed with saturated NaHCO₃ solution, subsequently with water and birne, dried over MgSO₄ and evaporated to dryness. The residue was recrystallized from ethanol to afford 7a (79% yield) as brown solid; mp 123–125°C; ¹H NMR (DMSO-*d*₆): δ 2.82 (4H, d, CH₂), 3.72 (2H, s, CH₂), 4.22 (2H, t, CH), 4.62 (2H, s, OH), 6.70 (2H, s, ArH), 6.73 (2H, d, J = 9.0 Hz, ArH), 6.84 (2H, d, J = 9.1 Hz, ArH), 6.99 (4H, m, ArH), 7.00 (2H, s, ArH), 7.21 (2H, s, ArH), 7.32 (4H, d, J = 9.2 Hz, ArH); ¹³C NMR (DMSO-d₆): δ 38.1, 39.3, 42.1, 110.5, 117.2, 118.7, 125.4, 126.5, 127.8, 128.0, 128.9, 130.0, 132.0, 132.8, 140.3, 143.4, 156.2, 161.7; IR (KBr): 3390, 3037, 2972, 1609, 1470, 1030 cm⁻¹; MS: m/z 590 (M⁺). The other compounds **7b–i** were also prepared by the similar procedure.

4-[4-Hydroxy-3-(6-phenylbenzo[*d*]isoxazol-4-yl)benzyl]-2-(6-phenylbenzo[*d*]isoxazol-4-yl)phenol (8a). To a solution of 7a (5.9 g, 0.01 mol) in dry benzene (20 mL), DDQ (6.81 g, 0.03 mol), dissolved in dry benzene (20 mL), was added in portions. The mixture was heated to reflux and stirred for 5 h under nitrogen atmosphere. The precipitated DDQ-H₂ was filtered off and the filtrate was subjected to column chromatography on silica gel (60–120 mesh) to afford pure **8a** (78% yield) as orange solid; mp 158–160°C; ¹H NMR (DMSO-*d*₆): δ 3.99 (2H, s, CH₂), 4.65 (2H, s, OH), 6.90 (2H, s, ArH), 6.99 (2H, d, *J* = 8.9 Hz, ArH), 7.29 (2H, d, *J* = 8.9 Hz, ArH), 7.44– 7.50 (10H, m, ArH), 7.90 (2H, s, ArH), 8.20 (2H, s, ArH), 8.60 (2H, s, ArH); ¹³C NMR (DMSO-*d*₆): δ 40.7, 115.1, 117.0, 125.4, 127.0, 127.9, 128.3, 130.0, 133.2, 134.7, 137.4, 144.9, 145.2, 150.4, 151.2, 153.6, 159.3, 171.4; IR (KBr): υ 3344, 3062, 2972, 1609, 1470, 1030 cm $^{-1}$; MS: m/z 586 (M $^+$). Anal. calcd. for $C_{39}H_{26}N_2O_4$: C, 79.85; H, 4.47; N, 4.78. Found: C, 79.90; H, 4.39; N, 4.71. The other compounds **8b–i** were also prepared by the similar procedure.

2-[6-(4-Bromophenyl)benzo[*d*]isoxazol-4-yl]-4-3-[6-(4-bromophenyl)benzo[*d*]isoxazol-4-yl]-4-hydroxybenzylphenol (**8b**). This compound was obtained as brown solid; yield 84%; mp 171–172°C; ¹H NMR (DMSO-*d*₆): δ 4.00 (2H, s, CH₂), 4.65 (2H, s, OH), 6.90 (2H, s, ArH), 7.00 (2H, d, *J* = 8.9 Hz, ArH), 7.29 (2H, d, *J* = 8.9 Hz, ArH), 7.40–7.45 (8H, m, ArH), 7.90 (2H, s, ArH), 8.20 (2H, s, ArH), 8.60 (2H, s, ArH); ¹³C NMR (DMSO-*d*₆): δ 40.7, 115.0, 117.1, 122.1, 125.4, 130.0, 130.6, 132.3, 133.1, 134.6, 137.3, 141.3, 144.9, 147.1, 151.3, 153.6, 159.4, 171.4; IR (KBr): υ 3390, 3065, 2995, 1609, 1470, 1030, 586 cm⁻¹; MS: *m/z* 742/744/746 (M⁺). Anal. calcd. for C₃₉H₂₄Br₂N₂O₄: C, 62.92; H, 3.25; N, 3.76. Found: C, 62.85; H, 3.30; N, 3.70.

2-[6-(4-Aminophenyl)benzo[*d*]isoxazol-4-yl]-4-3-[6-(4-aminophenyl)benzo[*d*]isoxazol-4-yl]-4-hydroxybenzylphenol (8c). This compound was obtained as brown solid; yield 79%; mp 158–160°C; ¹H NMR (DMSO-*d*₆): δ 4.06 (2H, s, CH₂), 4.65 (2H, s, OH), 6.68 (4H, d, *J* = 8.5 Hz, ArH), 6.90 (2H, s, ArH), 7.00 (2H, d, *J* = 8.9 Hz, ArH), 7.31 (2H, d, *J* = 8.9 Hz, ArH), 7.00 (2H, s, ArH), 8.20 (2H, s, ArH), 8.60 (2H, s, ArH); ¹³C NMR (DMSO-*d*₆): δ 40.7, 115.9, 116.7, 125.3, 127.3, 128.5, 130.1, 133.0, 134.1, 134.7, 137.5, 139.6, 144.5, 146.3, 151.2, 153.4, 159.2, 171.3; IR (KBr): υ 3390, 3065, 2972, 1612, 1469, 1028 cm⁻¹; MS: *m*/*z* 616 (M⁺). Anal. calcd. for C₃₉H₂₈N₄O₄: C, 75.96; H, 4.58; N, 9.09. Found: C, 75.85; H, 4.60; N, 9.03.

2-[6-(4-Chlorophenyl)benzo[*d*]isoxazol-4-yl]-4-3-[6-(4-chlorophenyl)benzo[*d*]isoxazol-4-yl]-4-hydroxybenzylphenol (8d). This compound was obtained as yellow solid; yield 86%; mp 144–146°C; ¹H NMR (DMSO-*d*₆): δ 4.02 (2H, s, CH₂), 4.68 (2H, s, OH), 6.90 (2H, s, ArH), 7.00 (2H, d, *J* = 8.9 Hz, ArH), 7.31 (2H, d, *J* = 8.9 Hz, ArH), 7.39 (4H, d, *J* = 8.1 Hz, ArH), 7.83 (4H, d, *J* = 8.1 Hz, ArH), 7.90 (2H, s, ArH), 8.51 (2H, s, ArH); ¹³C NMR (DMSO-*d*₆): δ 40.7, 115.7, 117.2, 125.4, 127.3, 128.5, 130.1, 133.0, 134.1, 134.7, 137.5, 139.6, 144.5, 146.3, 151.2, 153.4, 159.2, 171.3; IR (KBr): υ 3384, 3062, 2968, 1605, 1470, 1028, 782 cm⁻¹; MS: *m*/*z* 654/656/658 (M⁺). Anal. calcd. for C₃₉H₂₄Cl₂N₂O₄: C, 71.46; H, 3.69; N, 4.27. Found: C, 71.42; H, 3.61; N, 4.31.

4-4-Hydroxy-3-[6-(4-methoxyphenyl)benzo[*d***]isoxazol-4-yl]benzyl-2-[6-(4-methoxyphenyl)benzo[***d***]isoxazol-4-yl]phenol** (**8e**). This compound was obtained as yellow solid; yield 77%; mp 135–137°C; ¹H NMR (DMSO-*d*₆): δ 3.81 (6H, s, OMe), 4.02 (2H, s, CH₂), 4.70 (2H, s, OH), 6.90 (2H, s, ArH), 6.96 (4H, d, *J* = 8.4 Hz, ArH), 7.00 (2H, d, *J* = 8.9 Hz, ArH), 7.32 (2H, d, *J* = 8.9 Hz, ArH), 7.34 (4H, d, *J* = 8.4 Hz, ArH), 7.90 (2H, s, ArH), 8.22 (2H, s, ArH), 8.51 (2H, s, ArH); ¹³C NMR (DMSO-*d*₆): δ 40.5, 54.7, 115.0, 112.1, 117.0, 125.0, 129.5, 130.1, 133.1, 134.1, 137.3, 139.5, 144.3, 147.2, 151.1, 153.5, 159.1, 160.7, 172.1; IR (KBr): υ 3384, 3062, 2968, 1605, 1470, 1240 cm⁻¹; MS: *m*/*z* 646 (M⁺). Anal. calcd. for C₄₁H₃₀N₂O₆: C, 76.15; H, 4.68; N, 4.33. Found: C, 76.21; H, 4.61; N, 4.35.

2-[6-(2-Furyl)benzo[*d***]isoxazol-4-yl]-4-3-[6-(2-furyl)benzo-**[*d***]isoxazol-4-yl]-4-hydroxybenzylphenol (8f).** This compound was obtained as black solid; yield 80%; mp 162–163°C; ¹H NMR (DMSO- d_6): δ 4.02 (2H, s, CH₂), 4.70 (2H, s, OH), 6.34 (2H, m, ArH), 6.82 (2H, s, ArH), 6.70 (2H, m, ArH), 6.96 (2H, d, J = 8.9 Hz, ArH), 7.32 (2H, d, J = 8.9 Hz, ArH), 7.39 (2H, m, ArH), 7.90 (4H, m, ArH), 8.00 (2H, s, ArH), 8.22 (2H, s, ArH); ¹³C NMR (DMSO- d_6): δ 40.8, 106.2, 106.5, 114.4, 115.1, 121.7, 126.9, 134.1, 138.7, 139.6, 141.9, 143.1, 151.1, 159.1, 159.8, 176.4; IR (KBr): υ 3390, 3071, 2965, 1609, 1470, 1030 cm⁻¹; MS: m/z 566 (M⁺). Anal. calcd. for C₃₅H₂₂N₂O₆: C, 74.20; H, 3.91; N, 4.94. Found: C, 74.12; H, 3.90; N, 4.83.

4-4-Hydroxy-3-[6-(1,3-thiazol-5-yl)benzo[*d*]isoxazol-4-yl]benzyl-2-[6-(1,3-thiazol-5-yl)benzo[*d*]isoxazol-4-yl]phenol (8g). This compound was obtained as brown solid; yield 82%; mp 149– 150°C; ¹H NMR (DMSO-*d*₆): δ 4.02 (2H, s, CH₂), 4.70 (2H, s, OH), 6.74 (2H, s, ArH), 6.96 (2H, d, *J* = 8.9 Hz, ArH), 7.32 (2H, d, *J* = 8.9 Hz, ArH), 7.90–7.95 (4H, m, ArH), 8.10 (2H, s, ArH), 8.22 (2H, s, ArH), 8.62 (2H, s, ArH); ¹³C NMR (DMSO-*d*₆): δ 40.6, 105.9, 115.1, 120.1, 120.9, 126.3, 134.0, 137.4, 138.3, 138.9, 140.5, 142.7, 150.9, 151.7, 158.4, 160.3, 173.1; IR (KBr): υ 3390, 3071, 2965, 1609, 1580, 1470, 638 cm⁻¹; MS: *m/z* 600 (M⁺). Anal. calcd. for C₃₃H₂₀N₄O₄S₂: C, 65.99; H, 3.36; N, 9.33. Found: C, 65.89; H, 3.40; N, 9.24.

4-4-Hydroxy-3-[6-(2-pyridyl)benzo[*d*]isoxazol-4-yl]benzyl-**2-[6-(2-pyridyl)benzo**[*d*]isoxazol-4-yl]phenol (8h). This compound was obtained as brown solid; yield 77%; mp 151– 153° C; ¹H NMR (DMSO-*d*₆): δ 4.02 (2H, s, CH₂), 4.68 (2H, s, OH), 6.51 (2H, s, ArH), 7.00 (4H, m, ArH), 7.31 (2H, d, *J* = 8.9 Hz, ArH), 7.59 (2H, m, ArH), 7.72 (2H, m, ArH), 7.92 (2H, s, ArH), 8.22 (2H, s, ArH), 8.62 (2H, d, *J* = 4.7 Hz, ArH), 8.92 (2H, s, ArH); ¹³C NMR (DMSO-*d*₆): δ 40.8, 115.0, 117.0, 121.7, 124.7, 125.3, 130.1, 133.0, 134.0, 135.3, 137.1, 144.1, 144.8, 146.3, 151.1, 153.4, 157.9, 160.3, 169.4; IR (KBr): υ 3390, 3072, 2969, 1605, 1470 cm⁻¹; MS: *m/z* 588 (M⁺). Anal. calcd. for C₃₇H₂₄N₄O₄: C, 75.50; H, 4.11; N, 9.52. Found: C, 75.41; H, 4.15; N, 9.47.

4-4-Hydroxy-3-[6-(2-pyrazinyl)benzo[*d*]isoxazol-4-yl]benzyl-2-[6-(2-pyrazinyl)benzo[*d*]isoxazol-4-yl]phenol (8i). This compound was obtained as black solid; yield 81%; mp 162–164°C; ¹H NMR (DMSO-*d*₆): δ 4.02 (2H, s, CH₂), 4.68 (2H, s, OH), 6.72 (2H, s, ArH), 7.00 (2H, d, *J* = 8.9 Hz, ArH), 7.31 (2H, d, *J* = 8.9 Hz, ArH), 7.90 (2H, s, ArH), 8.22 (2H, s, ArH), 8.41 (2H, d, *J* = 2.7 Hz, ArH), 8.64 (2H, d, *J* = 2.7 Hz, ArH), 138.0, 142.1, 110.7, 115.1, 127.1, 130.4, 134.0, 137.1, 138.0, 142.1, 143.4, 143.9, 144.1, 144.9, 146.3, 151.2, 156.7, 158.4, 169.4; IR (KBr): υ 3384, 3062, 2968, 1609, 1470, 1028, cm⁻¹; MS: *m*/z 590 (M⁺). Anal. calcd. for C₃₅H₂₂N₆O₄: C, 71.18; H, 3.75; N, 14.23. Found: C, 71.11; H, 3.74; N, 14.17.

Acknowledgment. The authors are grateful to the Director, Indian Institute of Chemical Technology, Hyderabad, India, for NMR and MS spectral analysis and to the Head, Department of Biotechnology, S.R. PG. Center, Warangal, India for providing facilities for biological screening.

REFERENCES AND NOTES

[1] De Logu, A.; Saddi, M.; Cardia, M. C.; Borgna, R.; Sanna, C.; Saddi, B.; Maccioni, E. J Antimicrob Chemother 2005, 55, 692.

[2] (a) Desai, J. T.; Desai, C. K.; Desai, K. R. J Iran Chem Soc 2008, 5, 67; (b) Basawaraj, R.; Ahmad, A.; Omkar, K.; Sangapure, S. S. Indian J Heterocycl Chem 2007, 17, 11.

[3] Rajeshwar, N.; Marcus, P.; Stefanie, L.; Karlheinz, B.; Sabine, K.; Thomas, D.; Sascha, W.; Eckhard, M.; Boris, S. Chem Med Chem 2008, 3, 165.

[4] Cho, S. Y.; Ahn, J. H.; Ha, J. D.; Kang, S. K.; Back, J. Y.; Han, S. S.; Shin, E. Y.; Kim, S. S.; Kim, K. R.; Cheon, H. G.; Choi, J. K. Bull Korean Chem Soc 2003, 24, 1455.

[5] Schmidt, J.; Widmann, K. Lab Pure and Pharm; Lem Royal Technochschule: Stultgart Bet, 1908; Vol. 41, p 1252.

[6] Hansen, J. F.; Strong, S. A. J Heterocycl Chem 1977, 14, 1289.

[7] Tayade, V. B.; Jamode, V. S. Asian J Chem 1997, 9, 866.

[8] Balalaie, S.; Sharifi, A.; Ahangarian, B. Indian J Heterocycl Chem 2000, 10, 149.

[9] (a) Tibor, M.; Neil, P. S.; Henry Gount, S. P. PCT Int Appl WO 9414, 782; (b) Holla, B. S.; Ambekar, S. Y. J Indian Chem Soc 1973, 50, 673.

[10] Kachhadia, V. V.; Patel, M. R.; Joshi, H. S. J Sci Islamic Republic of Iran, 2004, 15, 47.

[11] Marcin, M.; Michal, Z.; Ewa, D. S.; Stanislaw, R. Cell Mol Biol Lett 2005, 10, 613.

[12] Nagar, N. R.; Shan, V. H.; Mehta, T. Indian J Heterocycl Chem 2003, 13, 173.

[13] Dannhardt, G.; Kiefer, W.; Kramer, G.; Maehrlein, S.; Nowe, U.; Fiebich, B. Eur J Med Chem 2000, 35, 499.

[14] (a) Gaziz, M. H.; Hosseir, N. L.; Doss, A. S.; Rock, S. Indian J Chem 1976, 14B, 286; (b) Witezak, A.; Mrolikowaska, M. Pol J Chem 1981, 55, 89; (c) Lohiya, S. B.; Ghiya, B. J. Indian J Chem 1986, 25B, 279.

[15] (a) Adam, G. M.; Christopher, J. E.; Stephen, F. L.; Gregory, W. S. J Org Chem 1998, 63, 9069; (b) Robert, E.; Sammelson, R.; Bryan, M.; Mark, J. K. J Org Chem 2000, 65, 2225; (c) Trond, V. H.; Peng, W.; Valery, V. F. J Org Chem 2005, 70, 7761.

[16] (a) Claisen, L. Ber 1903, 36, 3664; (b) Claisen, L. Ber 1909, 42, 59.

[17] Padmavathi, V.; Jagan Mohan Reddy, B.; Balaiah, A.; Venugopal Reddy, K.; Bhaskar Reddy, D. Molecules 2000, 5, 1281.

[18] (a) Smitha, G.; Sanjeeva Reddy, Ch. Synth Commun 2004, 34, 3997; (b) Smitha, G.; Sanjeeva Reddy, Ch. Synth Commun 2006, 36, 1795; (c) Sanjeeva Reddy, Ch.; Nagaraj, A. Heterocycl Commun 2007, 13, 67; (d) Sanjeeva Reddy, Ch.; Nagaraj, A. Chin Chem Lett 2007, 18, 1431; (e) Sanjeeva Reddy, Ch.; Nagaraj, A.; Jalapathi, P. Chin Chem Lett 2007, 18, 1213; (f) Nagaraj, A.; Sanjeeva Reddy, Ch. J Heterocycl Chem 2007, 44, 1357; (g) Nagaraj, A.; Sanjeeva Reddy, Ch. J Heterocycl Chem 2007, 44, 1181; (h) Sanjeeva Reddy, Ch.; Nagaraj, A. Indian J Chem 2008, 47B, 319; (i) Nagaraj, A.; Sanjeeva Reddy, Ch. J Iran Chem Soc 2008, 5, 262; (j) Srinivas, A.; Nagaraj, A.; Sanjeeva Reddy, Ch. Indian J Chem 2008, 47B, 787; (k) Srinivas, A.; Nagaraj, A.; Sanjeeva Reddy, Ch. J Heterocycl Chem 2008, 45, 999; (1) Srinivas, A.; Nagaraj, A.; Sanjeeva Reddy, Ch. J Heterocycl Chem 2008, 45, 1121; (m) Raghu, M.; Nagaraj, A.; Sanjeeva Reddy, Ch. J Heterocycl Chem 2008, 45, 1115; (n) Sanjeeva Reddy, Ch.; Nagaraj, A. Indian J Chem 2008, 47B, 1154.

[19] Marvel, C. S.; Tarkoy, N. J Am Chem Soc 1957, 79, 6000.

[20] National Committee for Clinical Laboratory Standards (NCCLs). Standard Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria, Which Grows Aerobically; National Committee for Clinical Laboratory Standards: Villanova, 1982; p 242.